

## The Bonding in XeF<sub>2</sub>

By R. C. CATTON and K. A. R. MITCHELL\*

(Department of Chemistry, University of British Columbia, Vancouver 8, British Columbia, Canada)

**Summary** A configuration-interaction calculation for XeF<sub>2</sub> shows that the valence-bond structures which incorporate the 5*d*<sub>z<sup>2</sup></sub> orbital at xenon contribute approximately 69% of the total wave function and that the resonance of F<sup>-</sup>Xe<sup>+</sup>F<sup>-</sup> and F<sup>-</sup>Xe<sup>+</sup>-F based on the 5*p*<sub>z</sub> orbital at xenon contributes only about 16% of the total wave function.

THE bonding in XeF<sub>2</sub> and in related molecules has been discussed qualitatively both with models which maintain

and F<sup>-</sup>Xe<sup>+</sup>-F, where the covalent bonds involve the overlaps of *h*<sub>1</sub> with 2*p*<sub>σ1</sub> (the 2*p*<sub>σ</sub> orbital at the fluorine atom F-1) and *h*<sub>2</sub> with 2*p*<sub>σ2</sub>;  $\psi_{\text{III}}$  which describes the purely ionic structure F<sup>-</sup>Xe<sup>2+</sup>F<sup>-</sup>; and  $\psi_{\text{IV}}$  which corresponds to the ionic-covalent resonance where the covalent bonds involve the overlaps of the 5*p*<sub>z</sub> orbital at Xe. The differences in the present work from the earlier study<sup>5</sup> are listed in the following: (i) The orbital 2*p*<sub>σ1</sub> is orthogonalised to the 5*s* orbital at Xe and to the 2*p*<sub>σ2</sub> orbital. (ii) A parameter *k* is introduced which governs the ionic character of the covalent

Energies of valence-bond structures and results of configuration interaction calculation

	$\psi_{\text{I}}$	$\psi_{\text{II}}$	$\psi_{\text{III}}$	$\psi_{\text{IV}}$	$\psi_{10c}$	$\psi$
Energies*	-1.99	-2.03	-1.66	-1.92	-2.31	-2.33
Coefficients in (1)	0.481	0.329	0.246	0.209	—	—
% Contributions to $\psi$	38.9	29.7	15.1	16.2	—	—

\* In atomic units (1 a.u. = 27.205 eV) and relative to the energy of the F<sup>+</sup>Xe<sup>2+</sup>F<sup>+</sup> core.

the octet at xenon<sup>1,2</sup> and with the model that emphasises the participation of outer-*d* orbitals.<sup>3,4</sup> We now report an extension of an earlier study<sup>5</sup> which attempted to provide a quantitative basis for discussions of the bonding in XeF<sub>2</sub>. The calculations use the valence-bond method and are subject to the following approximations: (i) Slater-type radial functions are used for each atomic orbital and the exponents are fixed for all orbitals other than the 5*d*<sub>z<sup>2</sup></sub> orbital at xenon. (ii) A constant F<sup>+</sup>Xe<sup>2+</sup>F<sup>+</sup> core is assumed for the different valence-bond structures. (iii) The electron-repulsion integrals are simplified with the Mulliken approximation. (iv) Bonding by *d*<sub>π</sub> orbitals is neglected.

The present discussion centres on four valence-bond wave functions namely,  $\psi_{\text{I}}$  corresponding to the structure F-Xe-F, which implies localised covalent bonds based on digonal *pd* hybrids (*h*<sub>1</sub> and *h*<sub>2</sub>) at Xe;  $\psi_{\text{II}}$  which corresponds to the resonance of the ionic-covalent structures F<sup>-</sup>Xe<sup>+</sup>F<sup>-</sup>

bonds. This parameter is specified by reference to a new wave function  $\psi_{10c}$  which derives from  $\psi_{\text{I}}$  by replacing *h*<sub>1</sub> by N(*h*<sub>1</sub> + *k* 2*p*<sub>σ1</sub>) and *h*<sub>2</sub> by N(*h*<sub>2</sub> + *k* 2*p*<sub>σ2</sub>), where N is the normalisation constant. (iii) The resonance between the ionic-covalent structures is now included; in  $\psi_{\text{IV}}$  the 5*p*<sub>z</sub> orbital of xenon is orthogonalised to the 1*s* and 2*s* orbitals at the fluorines. (iv) The secular equation is solved for the free mixing of the four valence bond wave functions as in equation (1).

$$\psi = C_{\text{I}}\psi_{\text{I}} + C_{\text{II}}\psi_{\text{II}} + C_{\text{III}}\psi_{\text{III}} + C_{\text{IV}}\psi_{\text{IV}} \quad (1)$$

The energy of  $\psi_{10c}$  is minimised with *k* equal to 0.52 and the 5*d*<sub>z<sup>2</sup></sub> exponent equal to 2.07. This latter value has been used in the configuration-interaction calculation. Estimates of the relative contributions to  $\psi$  of  $\psi_{\text{I}}$ ,  $\psi_{\text{II}}$ ,  $\psi_{\text{III}}$ , and  $\psi_{\text{IV}}$  have been made following the population analysis for

molecular orbitals in the LCAO approximation.<sup>6</sup> Thus, in the present context, the contribution of  $\psi_I$  to  $\psi$  is

$$C_I^2 + C_I C_{II} \langle \psi_I | \psi_{II} \rangle + C_I C_{III} \langle \psi_I | \psi_{III} \rangle + C_I C_{IV} \langle \psi_I | \psi_{IV} \rangle$$

The results in the Table bring out the importance of the valence-bond structures which involve the  $5d_{z^2}$  orbital at xenon in covalent bonding. The close approach of the

energy of  $\psi_{10c}$  to that from the configuration interaction calculation suggests the localised electron-pair model is probably reasonable for qualitative discussions. The analysis that expands  $\psi$  in terms of  $\psi_{10c}$  and the functions that include non-localised contributions shows that  $\psi_{10c}$  contributes 71% of the total wave function.

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